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Cyclic Dipeptides. A Stereocontrolled Synthesis of (2S,3R,6R)- and (2R,3R,6R)-6-tert-Butoxycarbonylamino-3-methoxycarbonyl-2-methyl-5-oxoperhydro-1,4-thiazepine

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Abstract: A stereselective synthesis of the title compounds, starting from commercially available amino acids, is described. The absolute stereochemistry of 3a and 3b has been deduced on the basis of ¹H NMR and chemical degradation studies. The formation of only these two isomers has been rationalized in terms of molecular mechanics calculations.

Cyclic dipeptides like 1 have recently attracted much interest as conformationally restricted dipeptide mimetics useful for the study of the binding conformation of enzyme substrates and receptor ligands. Moreover, structurally related macrocyclic sulphides and disulphides (e.g. 2) have been found to show pharmacological activities, such as hepatoprotective and immunomodulatory effects.

Being involved in the chemistry and the biological evaluation of sulphur containing cyclic compounds,³ and in the light of the importance of the biological activities cited above, we are reporting here the synthesis of enantiomerically pure perhydro-1,4-thiazepinecarboxylic acid derivatives 3, which can be regarded as relatives of 1 and 2.

Reaction between D-threonine methyl ester⁴ and S-acetyl-N-tert-butoxycarbonyl-L-cysteine 4 (prepared⁵ from N-tert-butoxycarbonyl-L-cystine) in the presence of N-cyclohexyl-N'-(2-morpholinoethyl)carbodiimide methyl p-toluenesulphonate (CMC) afforded after chromatographic purification the amide 5a ([α] $_D$ = +25, c=1, CHCl3) in 53% yield as a single stereoisomer as determined by TLC and ¹H NMR spectroscopy (Scheme 1). Treatment of 5a with methanesulphonyl chloride and diisopropylethylamine (DIPEA) at -5 °C gave the corresponding mesylate 6a exclusively as indicated by TLC and ¹H NMR spectroscopy. 6a proved to be unstable under the conditions of silica gel chromatography and was therefore used in the next reaction step without further purification. First attempts to convert 6a into the cyclic derivative 3a with methanolic ammonia or sodium borohydride led to a complex mixture of products from which 3a was isolated in only very small

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quantities after laborious chromatographic separations. Subsequently the deacetylation-cyclization reaction was performed with lithium trimethoxyaluminum hydride (3 molar equivalents) in dry THF. Under these conditions compounds 3a and 3b were obtained in 15% and 35% yield, respectively, along with the olefins 7 (E/Z mixture, 20%) and the disulphides 8 (12%). Bubbling nitrogen into the reaction mixture suppressed the formation of 8 and led to compounds 3a and 3b in a yield of 14% and 56%, respectively, while the olefins 7 were always present in trace amounts.

Scheme 1. Reagents: i) D-threonine methyl ester, CMC, CH₂Cl₂; ii) MsCl, DIPEA, CH₂Cl₂; iii) LiAlH(OMe)₃, THF.

The structures of 3a and 3b were determined by FAB-MS and 1H NMR spectroscopy.⁶ The stereochemistry of the methyl at C-2 and the methoxycarbonyl group at C-3 was proven to be *cis* for 3a by a nuclear Overhauser effect (NOE) between the C-2 and C-3 protons and *trans* in the case of 3b (where no NOE was detected). In order to establish the absolute stereochemistry at C-3, 3a was desulphurated with Ra-Ni to compound 9, which proved to be identical in all the respects, including specific rotation ($[\alpha]_D = -25$, c=2, CHCl₃), with the material obtained from commercially available BOC-L-alanine and (S)-(+)-2-aminobutyric acid (Scheme 2).

Scheme 2. Reagents: i) Ra-Ni, MeOH; ii) CH₂N₂; iii) CMC, CH₂Cl₂.

It is interesting to note that both 3a and 3b possess a C-3 stereochemistry inverted with respect to their precursor 6a. In order to obtain further insight into the reasons for the exclusive formation of only these two isomers 3a and 3b, we resorted to molecular mechanics calculations with the programme PCMODEL (MMX force field).⁷

All possible stereoisomers⁸ of 6-tert-butoxycarbonylamino-3-methoxycarbonyl-2-methyl-5-oxoperhydro-1,4-thiazepine were taken into account for the computational study and, after a preliminary geometry optimization, submitted to a statistical conformational search with BKM.⁹ The lowest energy conformers of stereoisomers 3a and 3b were found to be the most stable (Table 1).

Table 1. Energy (kcal/mol) of the lowest energy conformer of the four stereoisomers of 3.

BOCHN S 2 Me 3 CO ₂ Me	3a	3b	3a'	3b'
	(2S,3R)	(2R,3R)	(2R,3S)	(2S,3S)
)-N 3	E = 12.56	E = 12.57	E = 14.13	E = 14.33

In the light of these results, we can infer that the cyclization probably occurs through a preliminary methanesulphonic acid elimination from 6a, followed by Michael addition of the thiolate group giving rise to the thermodynamically more stable isomers (Scheme 3) In agreement with this hypothesis, the olefins 7 were isolated and submitted to the cyclizations under the same experimental conditions as used for 6a, leading indeed to the same products 3a and 3b.

Scheme 3. Mechanism for the formation of the cyclic dipeptides 3a, 3b.

In order to confirm this reaction pathway, the sequence was repeated starting from 4 and L-threonine methyl ester to give the intermediate 5b ($[\alpha]_D = -10$, c=1, CHCl3), which was in turn subjected to mesylation

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and cyclization as described for 5a. As expected the same diastereoisomers 3a and 3b were obtained (51% and 25%, respectively), although the ratio of 3a: 3b (2:1) in this case was inverted with respect to the previous reaction (1:4).

In summary, the above studies have afforded a stereoselective synthesis of only two (3a and 3b) of the four possible⁸ stereoisomeric 6-tert-butoxycarbonylamino-3-methoxycarbonyl-2-methyl-5-oxoperhydro-1,4-thiazepines in enantiomerically pure form starting from either D- or L-threonine, with the product ratios depending on the absolute configurations of the starting amino acids.

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References and Notes

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- 8. Assuming that the *L*-cysteinyl moiety retains its configuration throughout the synthetic pathway, four stereoisomers of 3 are possible.
- 9. Professor K. Steliou, Boston University, MA, U.S.A.

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